Received 1 June 2006

Accepted 6 June 2006

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Yong Ling,^a Hao Xu,^a Zhi-Hong Zou^b and Cheng Yao^a*

^aDepartment of Applied Chemistry, College of Science, Nanjing University of Technology, Nanjing 210009, People's Republic of China, and ^bDepartment of Medical Engineering, College of Chemistry & Chemical Engineering, Southeast University, Nanjing 210009, People's Republic of China

Correspondence e-mail: yaocheng@njut.edu.cn

Key indicators

Single-crystal X-ray study T = 296 KMean $\sigma(\text{C-C}) = 0.004 \text{ Å}$ R factor = 0.053 wR factor = 0.141 Data-to-parameter ratio = 15.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-(2-Chloroacetyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline

In the molecule of the title compound, $C_{13}H_{16}CINO_3$, the Ncontaining ring is not planar and has a flattened boat form. Intermolecular $C-H \cdot \cdot \cdot O$ hydrogen bonds link the molecules to form infinite chains and may be effective in the stabilization of the crystal structure.

Comment

Some of the substituted tetrahydroisoquinolines are known to produce a variety of pharmacological and biochemical actions on the adrenergic nervous system (Smissman *et al.*, 1976). The pharmacological effects of tetrahydroisoquinolines include lypolytic (Shonk *et al.*, 1971), bronchial relaxant (Miller *et al.*, 1975) and hypotensive activities (Holtz *et al.*, 1964). In recent years, recognition of the importance of tetrahydroisoquinolines as antihypertensive or anti-arrhythmic agents has resulted in increased interest in related compounds (Harrold *et al.*, 1988). Tetrahydroisoquinoline intermediates have been prepared in order to search for novel biological activity acting on calcium or potassium channels (Dai *et al.*, 1996). We report here the crystal structure of the title compound, (I), which is a substituted tetrahydroisoquinoline.



In the molecule of (I) (Fig. 1), the bond lengths and angles are within normal ranges (Allen *et al.*, 1987). Ring A (N/C6– C10) is not planar, having a total puckering amplitude $Q_{\rm T}$ = 1.016 (5) Å and a flattened boat form [$\varphi = -0.59$ (4)° and $\theta =$ 39.77 (3)°] (Cremer & Pople, 1975). Ring A has a pseudomirror plane passing through atoms C8 and C10, as can be deduced from the torsion angles (Table 1). Ring B (C3–C6/ C10/C11) is, of course, planar.

As can be seen from the packing diagram (Fig. 2), intermolecular $C-H\cdots O$ hydrogen bonds (Table 2) link the molecules to form infinite chains, in which they may be effective in the stabilization of the crystal structure. Dipoledipole and van der Waals interactions are also effective in the molecular packing.

Experimental

© 2006 International Union of Crystallography All rights reserved The title compound, (I), was prepared from a mixture of 6,7dimethoxy-1,2,3,4-tetrahydroisoquinoline (6.0 g, 31 mmol), synthe-

organic papers



Figure 1

The molecular structure of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

sized through the Pictet–Spengler reaction (Smissman *et al.*, 1976), and ClCH₂COCl (25.5 g, 226 mmol) stirred in CH₂Cl₂ (100 ml) in an ice-bath for 3 h. Saturated aqueous NaHCO₃ solution (100 ml) was added to this mixture. The layers were separated, and the aqueous layer was further extracted with CH₂Cl₂ (50 ml). The combined organic phases were washed with brine (100 ml), dried (MgSO₄), and concentrated *in vacuo*. The flaxen crude product was recrystallized from EtOH and afforded (I) as a white solid (yield 7.3 g, 87.4%; m.p. 386 K). Crystals were obtained by dissolving the white solid (0.3 g) in AcOEt–EtOH (2:1, 20 ml) and evaporating the solvents slowly at room temperature for about 15 d.

Z = 4

 $D_x = 1.405 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

Block, colourless

 $0.30\,\times\,0.20\,\times\,0.10$ mm

3 standard reflections

every 200 reflections

intensity decay: 1%

 $w = 1/[\sigma^2(F_o^2) + (0.06P)^2]$

where $P = (F_0^2 + 2F_c^2)/3$

+ 0.34P]

 $(\Delta/\sigma)_{\rm max} < 0.001$

 $\Delta \rho_{\rm max} = 0.28 \text{ e} \text{ Å}^{-1}$

 $\Delta \rho_{\rm min} = -0.24 \text{ e } \text{\AA}^{-3}$

2496 independent reflections

1541 reflections with $I > 2\sigma(I)$

 $\mu = 0.30 \text{ mm}^{-1}$

T = 296 (2) K

 $R_{\rm int}=0.023$

 $\theta_{\rm max} = 26.0^{\circ}$

Crystal data

 $\begin{array}{l} C_{13}H_{16}\text{CINO}_{3} \\ M_{r} = 269.72 \\ \text{Monoclinic, } P_{2_{1}}/a \\ a = 8.6561 \ (9) \\ \text{Å} \\ b = 17.7944 \ (18) \\ \text{Å} \\ c = 9.3090 \ (11) \\ \text{Å} \\ \beta = 117.22 \ (3)^{\circ} \\ V = 1275.1 \ (4) \\ \text{Å}^{3} \end{array}$

Data collection

Enraf-Nonius CAD4 diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (North *et al.*, 1968) $T_{\min} = 0.916, T_{\max} = 0.971$ 2662 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.053$ $wR(F^2) = 0.141$ S = 1.032496 reflections 163 parameters H-atom parameters constrained

Table 1

Selected torsion angles (°).

| C10-C6-C7-C8 | 27.9 (4) | C8-N-C9-C10 | -28.0(4) |
|--------------|-----------|--------------|----------|
| C9-N-C8-C7 | 58.7 (3) | C7-C6-C10-C9 | 1.9 (4) |
| C6-C7-C8-N | -56.4 (3) | N-C9-C10-C6 | -3.1 (4) |



Figure 2

A packing diagram for (I). Hydrogen bonds are shown as dashed lines.

Table 2

Hydrogen-bond geometry (Å, °).

| $D - H \cdots A$ | D-H | $H \cdot \cdot \cdot A$ | $D \cdot \cdot \cdot A$ | $D - \mathbf{H} \cdot \cdot \cdot A$ |
|-------------------------------------|-------------|-------------------------|-------------------------|--------------------------------------|
| $C1 - H1C \cdot \cdot \cdot O3^{i}$ | 0.96 | 2.57 | 3.473 (5) | 156 |
| $C2-H2C\cdots O3^{ii}$ | 0.96 | 2.58 | 3.308 (4) | 133 |
| $C8 - H8B \cdots O2^{iii}$ | 0.97 | 2.53 | 3.431 (4) | 155 |
| $C13-H13B\cdotsO1^{iii}$ | 0.97 | 2.50 | 3.364 (4) | 148 |
| Symmetry codes: (i) | -x + 1, -y, | -z + 1; (ii) | -x + 2, -y, - | z + 2; (iii) |

 $-x + \frac{3}{2}, y + \frac{1}{2}, -z + 1.$

H atoms were positioned geometrically, with C–H = 0.93, 0.96 and 0.97 Å for aromatic, methylene and methyl H, respectively, and constrained to ride on their parent atoms, with $U_{iso}(H) = xU_{eq}(C)$, where x = 1.5 for methyl H and x = 1.2 for all other H.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2000); software used to prepare material for publication: *SHELXTL*.

The authors thank the Centre for Testing and Analysis, Nanjing University, for support.

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